

A. K. Bauri,<sup>a</sup> Sabine Foro,<sup>b</sup>  
Hans-Jörg Lindner<sup>b</sup> and  
Sandip K. Nayak<sup>c\*</sup><sup>a</sup>Bioorganic Division, Bhabha Atomic Research  
Centre, Trombay, Mumbai 400 085, India,<sup>b</sup>Clemens Schöpf-Institut für Organische Chemie  
und Biochemie, Technische Universität  
Darmstadt, Petersenstrasse 22, D-64287  
Darmstadt, Germany, and <sup>c</sup>Bio-organic  
Division, Bhabha Atomic Research Centre,  
Trombay, Mumbai 400 085, IndiaCorrespondence e-mail:  
sknayak@magnum.barc.ernet.in

## Key indicators

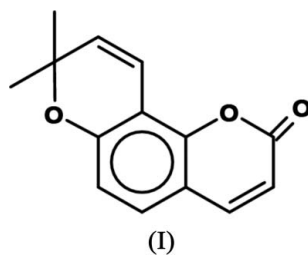
Single-crystal X-ray study  
 $T = 299$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å  
 $R$  factor = 0.045  
 $wR$  factor = 0.119  
Data-to-parameter ratio = 10.5For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.

## Reinvestigation of seselin

The crystal structure of seselin (2',2'-dimethyl-3-pyrano[6,5:7,8]coumarin),  $\text{C}_{14}\text{H}_{12}\text{O}_3$ , a bioactive pyrenocoumarin, has been reinvestigated. The coumarin ring system is nearly planar and the  $\alpha$ -pyran ring adopts a distorted half-chair conformation. An intermolecular  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bond is observed.

## Comment

Seselin, a bioactive pyrenocoumarin (Shanghag *et al.*, 1964; Smith *et al.*, 1957; Stanley & Vannier, 1957; Huang *et al.*, 1994), has been isolated from a methanol extract of the seeds of *Trachyspermum stictocarpum* (Ajmoda in Hindi). There is no previous phytochemical report on this plant in the literature. Several earlier publications dealing with its isolation from different natural sources (Austin *et al.*, 1968; Tomer *et al.*, 1969), structure elucidation (Shanghag *et al.*, 1967), synthesis (Schroeder *et al.*, 1959) and bioactivity studies (Huang *et al.*, 1994; Smith *et al.*, 1957) have appeared. From this perspective, our discovery of *T. stictocarpum* as a rich source of medicinally important pyranocoumarin is significant. The compound is credited with various medicinal attributes such as vasodilatory (Shanghag *et al.*, 1967), antitumor and anti-HIV activities (Huang *et al.*, 1994). This paper deals with the reinvestigation and improved refinement of the crystal structure of this compound (Kato, 1970), isolated from *T. stictocarpum*.



As shown in Fig. 1, the coumarin ring system is nearly planar, the largest deviation from the least-squares mean plane being 0.036 (1) Å at C1, and the  $\alpha$ -pyran ring adopts a distorted half-chair conformation.

In the crystal structure, the molecules are linked through intermolecular  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bonds, as shown in the packing diagram (Fig. 2). Details of the hydrogen bonding are given in Table 1.

## Experimental

The title compound was isolated as a major product from a methanol extract of *Trachyspermum stictocarpum* by column chromatography

Received 13 December 2005  
Accepted 6 March 2006

over silica gel with gradient elution by changing the polarity of the ethyl acetate–petroleum ether solvent system. Crystals suitable for X-ray diffraction were obtained by recrystallization from benzene–hexane (1:4) at room temperature by evaporation.

#### Crystal data

$C_{14}H_{12}O_3$   
 $M_r = 228.24$   
 Monoclinic,  $P2_1/c$   
 $a = 8.428$  (1) Å  
 $b = 11.112$  (2) Å  
 $c = 12.324$  (2) Å  
 $\beta = 103.08$  (1)°  
 $V = 1124.2$  (3) Å<sup>3</sup>  
 $Z = 4$

$D_x = 1.348$  Mg m<sup>-3</sup>  
 Cu  $K\alpha$  radiation  
 Cell parameters from 25 reflections  
 $\theta = 5.4$ – $27.7^\circ$   
 $\mu = 0.78$  mm<sup>-1</sup>  
 $T = 299$  (2) K  
 Prism, light yellow  
 $0.80 \times 0.38 \times 0.30$  mm

#### Data collection

Enraf–Nonius CAD-4 diffractometer  
 $\omega/2\theta$  scans  
 Absorption correction: none  
 3435 measured reflections  
 2004 independent reflections  
 1834 reflections with  $I > 2\sigma(I)$   
 $R_{int} = 0.023$

$\theta_{max} = 66.9^\circ$   
 $h = -9 \rightarrow 10$   
 $k = -13 \rightarrow 7$   
 $l = -14 \rightarrow 0$   
 3 standard reflections  
 frequency: 120 min  
 intensity decay: 0.9%

#### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.045$   
 $wR(F^2) = 0.119$   
 $S = 1.12$   
 2004 reflections  
 191 parameters  
 Only H-atom coordinates refined

$w = 1/[\sigma^2(F_o^2) + (0.0668P)^2 + 0.1549P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{max} = 0.002$   
 $\Delta\rho_{max} = 0.16$  e Å<sup>-3</sup>  
 $\Delta\rho_{min} = -0.27$  e Å<sup>-3</sup>  
 Extinction correction: *SHELXL97*  
 Extinction coefficient: 0.082 (4)

**Table 1**

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$C5-H5\cdots O3^i$	0.947 (18)	2.400 (18)	3.2710 (17)	152.9 (14)

Symmetry code: (i)  $-x + 1, y - \frac{1}{2}, -z - \frac{1}{2}$ .

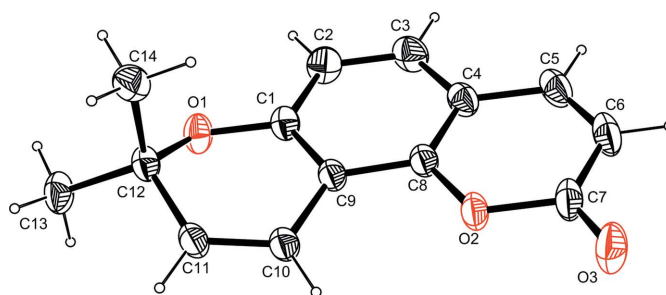
All H atoms were located in a difference map and refined with isotropic displacement parameters [ $1.2U_{eq}(\text{parent atom})$ ].

Data collection: *CAD-4/PC Software* (Enraf–Nonius, 1993); cell refinement: *CAD-4/PC Software*; data reduction: *REDU4* (Stoe & Cie, 1987); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPIII* (Burnett & Johnson, 1996), *ORTEP-3 for Windows* (Farrugia, 1997) and *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97*.

The authors thank Professor Dr Hartmut Fuess, FG Strukturforschung, FB Material- und Geowissenschaften, Technische Universität Darmstadt for diffractometer time.

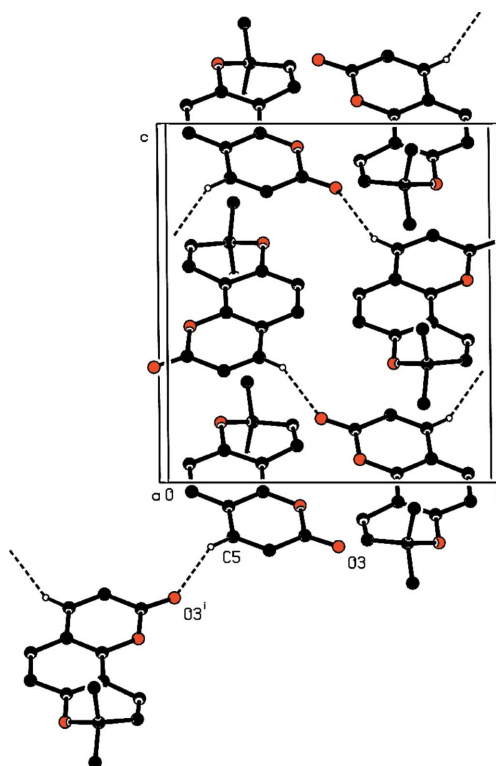
#### References

- Austin, P. W., Seshadri, T. R., Sood, M. S. & Vishwapaul (1968). *Tetrahedron*, **24**, 3247–3253.  
 Burnett, M. N. & Johnson, C. K. (1996). *ORTEPIII*. Report ORNL-6895. Oak Ridge National Laboratory, Tennessee, USA.  
 Enraf–Nonius (1993). *CAD-4/PC Software*. Version 1.2. Enraf–Nonius, Delft, The Netherlands.



**Figure 1**

The molecular structure of (I) with the atom-labeling scheme. Displacement ellipsoids are drawn at the 30% probability level.



**Figure 2**

The molecular packing of (I), with hydrogen bonds shown as dashed lines. H atoms not involved in hydrogen bonding have been omitted for clarity. [Symmetry code: (i)  $-x, y - \frac{1}{2}, -z - \frac{1}{2}$ ].

- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.  
 Huang, L., Kashiwada, Y., Cosentino, L. M., Fan, S., Chen, C.-H., McPhail, A. T., Fujioka, T., Mihashi, K. & Lee, K.-H. (1994). *J. Med. Chem.* **37**, 3947–3955.  
 Kato, K. V. (1970). *Acta Cryst.* **B26**, 2022–2029.  
 Schroeder, H. D., Benzze, W., Halpern, O. & Schmid, H. (1959). *Chem. Ber.* **92**, 2338–2363.  
 Shangha, S. N., Maheshwari, M. L. & Bhattacharyya, S. C. (1967). *Tetrahedron*, **23**, 1235–1240.  
 Shangha, S. N., Mesta, C. K., Maheshwari, M. L., Paknikar, S. K. & Bhattacharyya, S. C. (1964). *Tetrahedron*, **20**, 2605–2615.  
 Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.  
 Smith, E., Hosansky, N., Bywater, W. G. & Tamelen, E. E. van (1957). *J. Am. Chem. Soc.* **79**, 3534–3540.  
 Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.  
 Stanley, W. L. & Vannier, S. H. (1957). *J. Am. Chem. Soc.* **79**, 3488–3491.  
 Stoe & Cie (1987). *REDU4*. Version 6.2c. Stoe & Cie GmbH, Darmstadt, Germany.  
 Tomer, E., Goren R. & Monselise, S. P. (1969). *Phytochemistry*, **8**, 1315–1316.